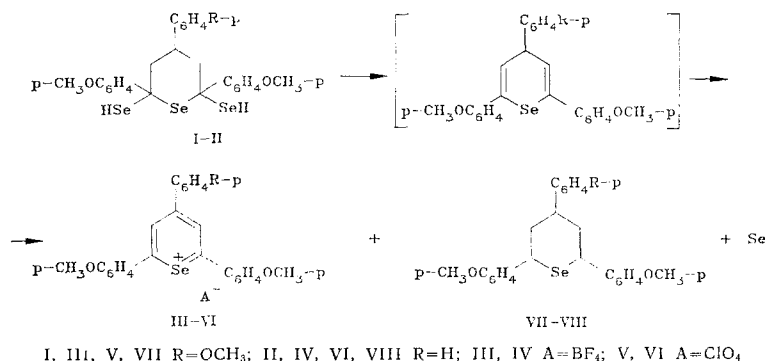


TRANSFORMATIONS OF 2,4,6-TRIARYL-2,6-DI(HYDROSELENO)-1-SELENOCYCLOHEXANES
ON THEIR REACTION WITH ACIDS

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UDC 547.818.9

Compounds of the 2,6-di(hydroseleno)-1-selenocyclohexane series have not previously been known, since their formation as stable structures under the action of hydrogen selenide on 1,5-diketones under the conditions of acid catalysis has proved unlikely in view of their semiacetal nature. However, it has been established that compounds (I) and (II) are formed under the conditions given above with high yields and are characterized by adequate stability [1]. Furthermore, we have proposed a method for obtaining selenopyrylium salts by the reaction of the compounds mentioned with acids in benzene solution [2]. On studying this reaction, we have established that, in addition to selenopyrylium salts, the corresponding selenocyclohexanes (VII) and (VIII) are formed. In this case, there is probably a disproportionation of the 4H-selenopyrans formed on the splitting out from compounds (I) and (II) of hydrogen selenide, which is oxidized by atmospheric oxygen to elementary selenium.



As the acid reagents we used BF₃·Et₂O, CF₃COOH, and 70% perchloric acid. When CF₃COOH was used, the resulting selenopyrylium trifluoroacetates were converted without isolation into the corresponding perchlorates (V) and (VI). When perchloric acid was used it was impossible to isolate the selenocyclohexanes (VII) and (VIII).

We give the compounds, mp, °C, yield, %, and acid used: (III), 244-245, 60, BF₃·Et₂O; (IV), 221-222, 59, BF₃·Et₂O; (V), 268-270, 53, CF₃COOH; (VI), 214-215, 56, CF₃COOH; (VII), 115-116, 30; (VIII), 124.5-126.30. The structures and compositions of the compounds obtained were confirmed by the results of elementary analysis and IR and PMR spectra.

LITERATURE CITED

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